B2, 1 Health and lifestyle	B2, 2 Life processes	B2, 3 Adaptation and inheritance
1.1 Nutrients - Name the nutrients required by the human body - Describe the components of a healthy diet - Explain the role of each nutrient in the body - Interpret nutritional information on food packaging to identify a healthy food 1.2 Food tests Key Practical - Describe how to test foods for starch, lipids, sugar, and protein - Describe the positive result for each food test - Use appropriate techniques to carry out a range of food tests safely - Explain why testing food is important	2.1 Photosynthesis Key Practical - State where photosynthesis occurs in a plant - State the word equation for photosynthesis - Carry out an experiment to test for the presence of starch in a leaf and recall the method - Explain how the plant obtains the reactants for photosynthesis 2.2 Leaves (practical) - Describe the structure and function of the main components of a leaf - Make observations of stomata from the underside of the leaf, and record observations as a labelled diagram - Explain how the structures of the leaf make it well adapted for photosynthesis	3.1 Competition and adaptation - State some resources that plants and animals compete for - Describe how organisms are adapted to their environments - Explain how adaptations help an organism survive in their environment 3.2 Natural Selection - State how adaptations affect survival rates - Describe the process of natural selection - Explain how natural selection leads to evolution
Digestive System Describe the structure and function of the main parts of the digestive system Describe the process of digestion Explain how each part of the digestive system works in sequence, including adaptations of the small intestine for its function	2.3 Aerobic respiration (practical) - State the word equation for aerobic respiration - Describe the process of respiration - Describe how to test for the products of aerobic respiration - Compare the process of respiration and combustion - Explain how the reactants for respiration get into the cells	3.3 Variation - State what is meant by the term variation - Describe the difference between environmental and inherited variation - Suggest examples of each type of variation - Record and categorise observations of variations and present findings graphically
 1.4 Enzymes in digestion Key Practical Describe the role of enzymes in digestion Record experimental data using a suitable results table and interpret results on enzymes Explain how enzymes affect the rate of digestion 	2.4 Anaerobic respiration - State the word equation for anaerobic respiration - Describe the differences between aerobic and anaerobic respiration - Explain the problems with anaerobic respiration - Discuss oxygen debt	3.4 Inheritance Key Practical - State what DNA is - State what is meant by a gene - Correctly follow a method to extract DNA - Describe how characteristics are inherited - Explain how characteristics are inherited through and coded for by genes
1.5 Alcohol, Smoking and Drugs - Describe the effect of alcohol and smoking on health and behaviour and on pregnancy - Explain how smoking causes disease - Describe the difference between recreational and medicinal drugs - Describe the effects of drugs on health and behaviour	Describe what food chains and food webs show Construct food chains for a specific ecosystem Identify prey/predator and producer/consumer in food webs Explain the link between food chains and energy	
	2.6 Disruptions to food chains and webs - State how one population of organisms can affect another - Describe the interdependence of organisms within a food web - Describe how toxic materials can accumulate in a food web	

C2, 1 The Periodic Table	C2, 2 Separation techniques	C2, 3 Metal reactions	C2, 4 The Earth
1.1 Metals and non-metals (practical)	2.1 Mixtures	3.1 Metals and oxygen (practical)	4.1 Earth & atmosphere
 State some common properties of metals and non-metals Use position on the Periodic Table to suggest if an element is a metal or a non-metal Use practical observations about materials to decide if they are metals or non-metals 	 Describe particle arrangements in mixtures Explain how to identify pure substances Select appropriate separation techniques for different mixtures Explain why separation techniques are suitable, in terms of the properties of constituent substances 	 Compare the reactions of different metals with oxygen Write word equations for the reaction of metals with oxygen Identify state symbols from an equation Construct balanced equations that include state symbols 	 Describe the layers of the Earth Describe the composition of the atmosphere Discuss how the atmosphere has changed over time
1.2 Groups and periods	2.2 Solutions	3.2 Metals and acid (practical)	4.2 The Carbon Cycle
 State what a group and period are Describe how the periodic table is organised Describe what pattern is shown in a given property of a group or period Use patterns to predict properties of elements 	 Identify a solvent, solute, and solution in a given scenario Define the key terms solvent, solute, solution, soluble and insoluble Use the particle model to explain dissolving Use data to predict how much solute is dissolved in a solution or the mass of a solution 	- Compare the reactions of different metals with dilute acids - Decide which metals react more vigorously from practical observations - Describe the test for hydrogen gas - Use word equations to show what happens when metals react with different acids	 Describe the key stages of the carbon cycle Identify which processes remove carbon dioxide from the atmosphere Describe different stores of carbon
1.3 The elements of Group 1 (practical demo)	2.3 Solubility key Practical Write up	3.3 Metal displacement reactions	4.3 Climate change
 Make observations about the reactivity of Group 1 metals in water Describe how different properties change for the elements of Group 1 Write word equations for the reaction of Group 1 metals with water Use patterns to predict properties of Group 1 elements Write balanced equations for the reaction of Group 1 metals with water 	- State what a saturated solution is - Describe the meaning of solubility - Describe how temperature affects solubility - Plan an investigation to compare solubility with temperature, considering variables - Explain why temperature affects the amount of solute dissolved in a solution	 State where different metals are found in the reactivity series Predict if a given pair of substances will undergo displacement Use the reactivity series to explain displacement reactions 	 Describe the greenhouse effect Explain why global warming happens Discuss some impacts of global warming Explain ways to reduce climate change
1.4 The elements of Group 7 (practical)	2.4 Filtration and Evaporation Key Practical	3.4 Extracting metals (practical)	4.4 Energy resources
 State key features of group 7 Use patterns to predict properties of Group 7 elements Write word equations to represent displacement reactions Explain displacement reactions 	 Describe how to filter a mixture Explain how to use evaporation to separate mixtures Plan a method to separate a mixture of rock salt Use particle diagrams to illustrate how filtering works 	 State where carbon is found in the reactivity series Use the reactivity series to decide which metals can be extracted from their ores by heating with carbon Explain why metals can be extracted using carbon, using the idea of displacement 	 Describe the difference between renewable and non-renewable Compare the advantages and disadvantages of using renewable and non-renewable

			- Describe how electricity is generated in a power station
1.5 The elements of Group 0	2.5 Chromatography Key Practical	3.5 Recycling	
 State properties of Group 0 elements Describe the reactivity of Group 0 elements Interpret data to plot a graph of atomic masses for Group 0 elements 	 Explain how chromatography separates mixtures Carry out a chromatography practical Analyse chromatograms to identify substances in mixtures Calculate the RF value on a chromatogram 	 Describe how aluminium is recycled Discuss the advantages and disadvantages of recycling Compare how other materials are recycled with the recycling of aluminium Explain why recycling metals reduces environmental impacts 	

P2, 1 Electricity and magnetism	P2, 2 Energy	P2, 3 Pressure
1.1 Static electricity	2.1 Energy transfers	3.1 Pressure in gases
 Explain how objects can become charged Describe how charged objects interact Describe what is meant by an electric field 	 Identify different forms of energy Describe simple energy transfers Discuss useful and wasted energy 	-Describe the factors that affect gas pressure Describe how atmospheric pressure changes with height
1.2 Electrical Circuits	2.2 Food and fuels Key Practical	P3.2 Pressure in liquids
 Identify common circuit symbols Use circuit symbols correctly to draw a circuit diagram Describe what is meant by current 	 Identify food and fuel as stores of energy Carry out an investigation to measure the energy in different food or fuels Complete a risk assessment identifying risks and precautions Use data gathered to compare fuels 	 Describe how liquid pressure changes with depth Explain why some things float and some things sink
1.3 Series Circuits Key Practical	2.3 Energy and temperature (practical)	P3.3 Pressure in solids
 Describe what a series circuit is Measure current in a series circuit Describe how current varies in a series circuit 	 State the difference between energy and temperature Describe what happens when you heat up solids, liquids, and gases Explain how a thermometer work in terms of expansion 	 recall the units of pressure Explain why pressure depends upon surface area Calculate pressure given relevant data by using an equation
1.4 Parallel Circuits Key Practical	2.4 Energy transfer: Conduction Key Practical	
 Describe what a parallel circuit is Describe the difference between series and parallel circuits Measure current in a parallel circuit Describe how current varies in a parallel circuit 	 Describe how energy is transferred by particles in conduction Discuss which materials are conductors or insulators Plan an investigation to compare different insulators 	-
1.5 Potential difference Key Practical	2.5 Energy transfer: Convection (practical)	
 Describe what is meant by potential difference Measure the potential difference in a series and parallel circuit Compare the potential difference in a series and parallel circuit 	Describe how energy is transferred in convection Describe how convection is used in everyday appliances Explain convection with reference to particles and density	-
1.6 Power	2.6 Energy transfer: Radiation Key Practical	-
 explain the difference between energy and power describe the link between power, fuel use, and the cost of using domestic appliances Use equations to complete calculations 	 Describe some sources of infrared radiation Explain how energy is transferred by radiation Carry out an investigation into which materials emit/absorb infrared radiation 	-
1.5 Magnets (practical)		

 State which materials are magnetic Describe features of a magnet Describe how magnets interact Discuss how to induce magnetism 	-	-
1.6 Magnetic fields (practical)		
 Draw the magnetic field lines around a bar magnet State the Earth has a magnetic field Explain how a compass works Describe two methods to plot a magnetic field 	-	-
1.7 Electromagnets Key Practical		
 Describe how to make an electromagnet Describe how to change the strength of an electromagnet Predict and test the effect of changes to an electromagnet Compare and contrast permanent and electromagnets 		